



education

DEPARTMENT: EDUCATION
MPUMALANGA PROVINCE

GRADE 12

PHYSICAL SCIENCES MONTHLY TEST

APRIL 2020

TOPIC: ORGANIC MOLECULES

MEMORANDUM

MARKS: 55

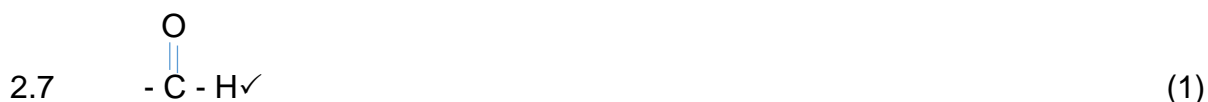
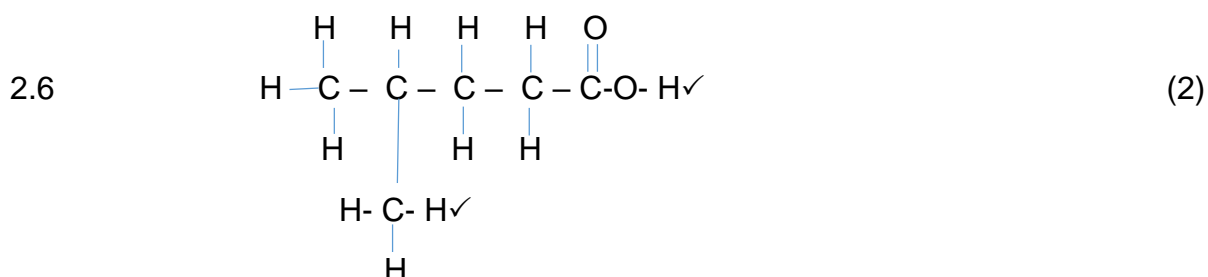
This memorandum consists of 4 pages

QUESTION 1

- 1.1 D✓✓ (2)
1.2 C✓✓ (2)
1.3 A✓✓ (2)
1.4 C✓✓ (2)
1.5 B✓✓ (2)
[10]

QUESTION 2

- 2.1 B✓✓ (2)
2.2 A✓ (1)
2.3 Chloroethene✓✓ (OR 1-Chloroethene) (2)
2.4 H✓ (1)
2.5 Carboxylic acid✓ (1)



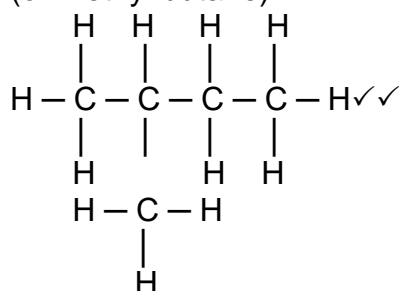
[10]

QUESTION 3

- 3.1.1 Boiling does not involve breaking the chemical (intra molecular) bonds between atoms.✓ OR Boiling involves overcoming the intermolecular forces between molecules. (1)
3.1.2 The longer chain in pentane provides greater surface area with stronger London forces. ✓
More energy is required to overcome the stronger forces.✓ (2)

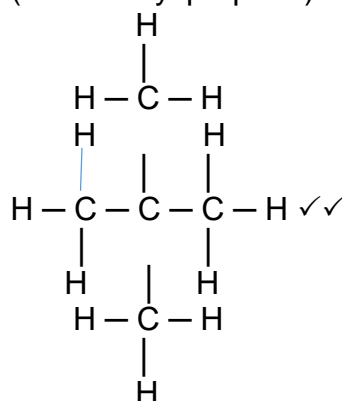
3.1.3 OPTION 1:

2-methylbutane✓
(or methyl butane)



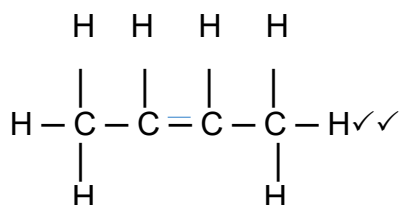
OPTION 2:

2,2-dimethylpropane✓
(or dimethyl propane)



(3)

3.2



(2)

3.3 A series of organic compounds that can be described by the same general formula OR in which one member differs from the next with a CH₂ group.✓✓ (2)

3.4.1 B✓ (1)

3.4.2 A✓ (1)

3.5 - increase in branching (in A) makes the molecule more spherical (compact) with less surface area over which the London forces work.✓

- This decreases the strength of London forces (in A). ✓

- Less energy is required to overcome the London forces (in A). ✓

- There will be more molecules/ higher vapour pressure above the surface of the substance (A). (3)

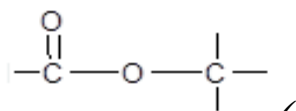
3.6.1 D (1)

3.6.2 C (1)

[17]

QUESTION 4

4.1.1 Esterification / Condensation ✓ (1)



4.1.2 ✓ (1)

4.1.3 Propanoic acid ✓ (1)

4.2.1

| Element | C | H | O |
|------------------------------|--|---|--|
| $n = \frac{m}{M} \checkmark$ | $\frac{3.758}{12} \checkmark$ = 0.313 | $\frac{0.316}{1} \checkmark$ = 0.316 | $\frac{1.251}{16} \checkmark$ = 0.078 |
| Divide by smallest | = 1 = 4 | = 1 = 4 | = 0,24 (x4) = 1 \checkmark |
| Empirical formula: | C ₄ H ₄ O \checkmark | | |

(6)

4.2.2

$$(C_4H_4O)_n = 136$$

$$(12 \times 4 + 1 \times 4 + 16)_n = 136$$

$$n = 2$$

$$\text{Molecular formula} = C_8H_8O_2 \checkmark \checkmark$$

(2)

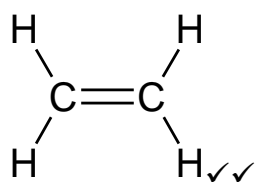
4.3.1 Dehydration / Elimination \checkmark

(1)

4.3.2 Concentrated sulphuric acid / H₂SO₄ / Phosphoric acid / H₂PO₄ \checkmark

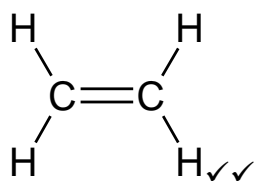
(1)

4.3.3



(2)

4.4.1



(2)

4.4.2 Addition \checkmark

(1)

[18]

TOTAL: 55