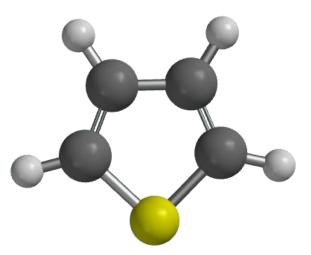
2302687 – Heterocyclic Compounds – Part I

Lecture 5-1

# **Ring Synthesis of Furan and Thiophene**



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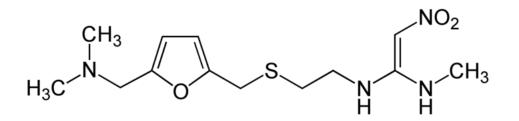
**Recommended Textbook:** 

Heterocyclic Chemistry, 5th Edition, J. A. Joule, K. Mills, 2010, Wiley

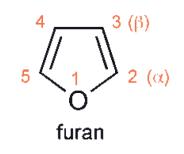
### **Furan**

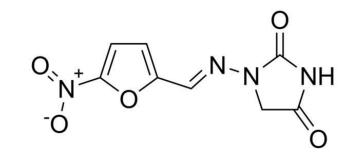
- Volatile, fairly stable compounds with pleasant odours
- Slightly soluble in water
- Precursor of the widely used solvent tetrahydrofuran (THF)
- Produced by the gas-phase decarbonylation of furfural

### **Drugs & Bioactive Furans**



**Ranitidine** is one of the biggest selling drugs in history; it is used to treat and prevent ulcers in the stomach and intestines





#### Name: Nitrofurantoin

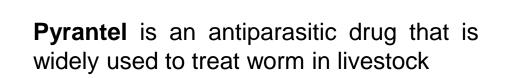
2008 Sales: \$164 million

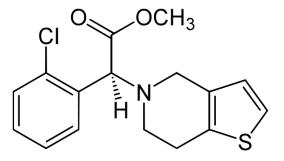
Disease: Antibiotic for urinary tract infections

# Thiophene

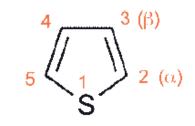
- Stable liquids that closely resemble the corresponding benzene compounds in boiling points and even in smell
- Occur in coal-tar distillates
- Thiophene is named after **theion** (the Greek word for sulfur) and **phaino** (the Greek word for shining)

### **Drugs & Bioactive Thiophenes**





Name: **Plavix** 2008 Sales: \$3.8 billion Company: Bristol-Myers Squibb Disease: Stroke and heart attack

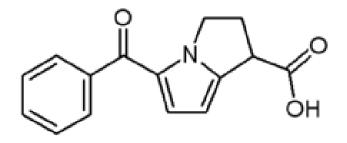


thiophene

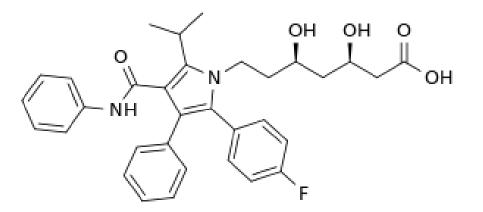
# Pyrrole

- Colourless liquids, with relatively weak odours rather like that of aniline, which, also like the anilines, darken by autoxidation
- Manufactured by alumina-catalysed gas-phase interaction of furan and ammonia
- First isolated from coal tar in 1834 and then in 1857 from the pyrolysate of bone

### **Drugs & Bioactive Pyrrole**



**Ketorolac** is a pain relief and anti-inflammatory drug

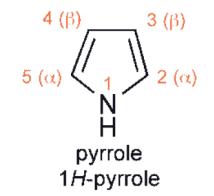


#### Name: Lipitor

2008 Sales: \$5.88 billion (Top-selling)

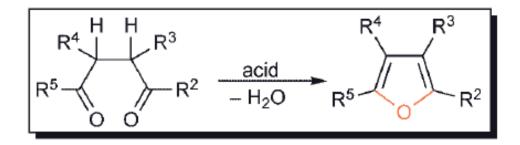
Company: Pfizer

Disease: Lowers LDL levels



## **Ring Synthesis of Furan**

1) From 1,4-Dicarbonyl Compounds



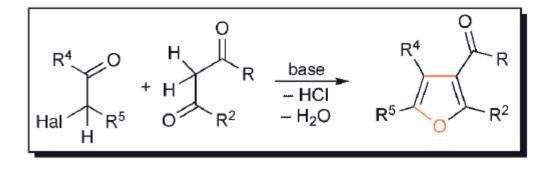
#### The Paal – Knorr Synthesis

- The most widely used approach to furans
- Usually, non-aqueous acidic conditions are employed to encourage the loss of water

$$t-Bu \xrightarrow{\mathsf{T}} t-Bu \xrightarrow{\mathsf{PhH}, \text{ reflux}}_{\mathsf{80\%}} \left[ t-Bu \xrightarrow{\mathsf{OH} O' \mathsf{H}^+}_{\mathsf{OH} O' \mathsf{H}^+} t-Bu \xrightarrow{\mathsf{T}}_{\mathsf{C}} t-Bu \xrightarrow{\mathsf{OH} O' \mathsf{H}^+}_{\mathsf{OH} O' \mathsf{H}^+} t-Bu \xrightarrow{\mathsf{C}}_{\mathsf{OH} O' \mathsf{OH} O' \mathsf{H}^+} t-Bu \xrightarrow{\mathsf{C}}_{\mathsf{OH} O' \mathsf{H}^+} t-Bu \xrightarrow{\mathsf{C}}_{\mathsf{OH} O' \mathsf{OH} O' \mathsf{H}^+} t-Bu \xrightarrow{\mathsf{C}}_{\mathsf{OH} O' \mathsf{OH} O' \mathsf{H}^+} t-Bu \xrightarrow{\mathsf{C}}_{\mathsf{OH} O' \mathsf{OH} O'$$

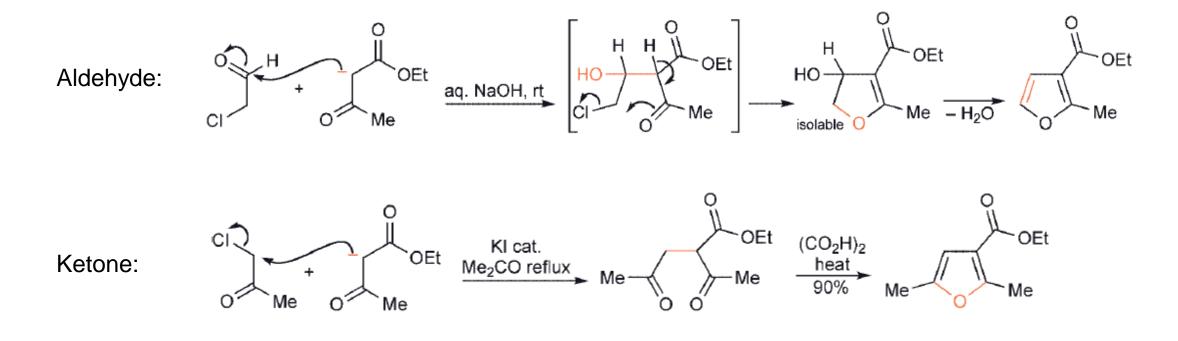
# **Ring Synthesis of Furan**

2) From *α*-Halo-Carbonyl and 1,3-Dicarbonyl Compounds



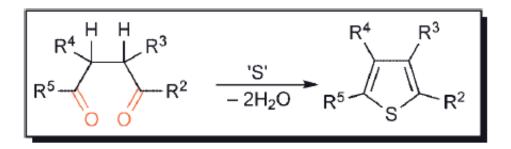
#### **The Feist–Benary Synthesis**

• Initial aldol condensation at the carbonyl carbon of a 2-halo-carbonyl then ring closure is achieved via intramolecular displacement of halide by enolate oxygen

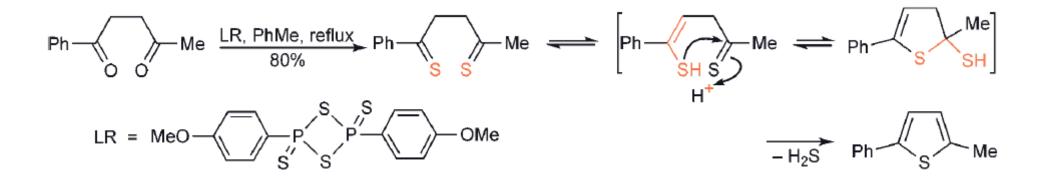


### **Ring Synthesis of Thiophene**

1) From 1,4-Dicarbonyl Compounds and a Source of Sulphide

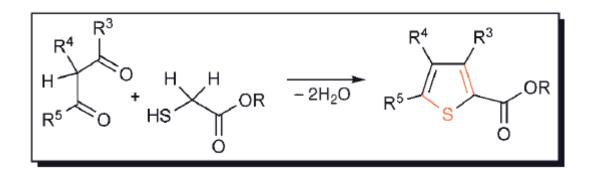


Sources of sulphide such as phosphorus sulphides, Lawesson's reagent (LR)

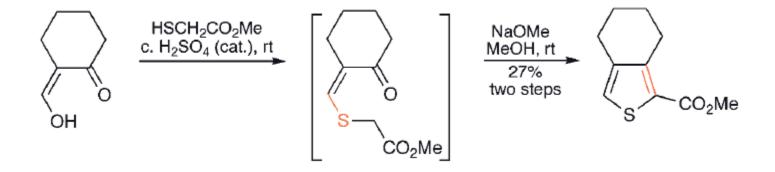


### **Ring Synthesis of Thiophene**

2) From Thioglycolates and 1,3-Dicarbonyl Compounds

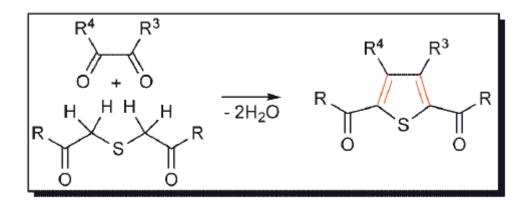


In most of the examples of this approach, thioglycolates, as donors of an S–C unit, have been reacted with 1,3-keto-aldehydes, to give intermediates that can be ring closed to give thiophenes



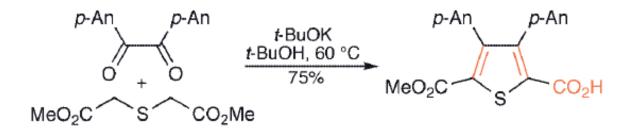
# **Ring Synthesis of Thiophene**

3) From Thiodiacetates and 1,2-Dicarbonyl Compounds



#### The Hinsberg Synthesis

• Two consecutive aldol condensations between a 1,2-dicarbonyl compound and diethyl thiodiacetates give thiophenes. The immediate product is an ester-acid.



#### **The Hinsberg Synthesis**

