2302687 – Heterocyclic Compounds – Part I Lecture 1-2



Nomenclature of Heterocyclic Compounds Part 1



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Recommended Textbook:

Heterocyclic Chemistry, 5th Edition, J. A. Joule, K. Mills, **2010**, Wiley

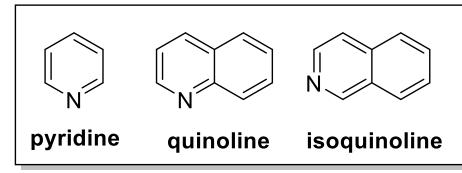
Nomenclature of Heterocyclic Compounds

- Early discovered heterocycles were received common/trivial names
- With millions of compounds to deal with, convenient, generally accepted systems must be available for the naming of the compounds
- Highly versatile system was created by A. Hantzsch in 1887 and independently by O. Widman in 1888 for the naming of 5- and 6-membered rings containing nitrogen
- The system was later applied to different ring sizes and to rings with other heteroatoms; it is now known as the Hantzsch-Widman system
- For instance; these 3 heterocyclics have common/trivial name in bold and systematic names beneath in *italic*

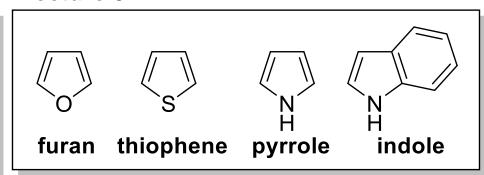


Common Names of Heterocyclic Compounds

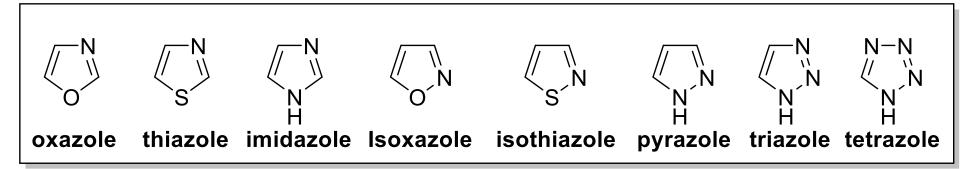
Lecture 4



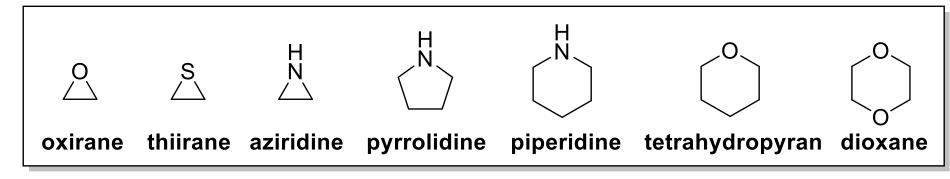
Lecture 5



Lecture 6



2nd half of the course



The systematic names from the **Hantzsch–Widman naming system** can be derived using the following **four rules**:

Rule #1 The heteroatom is given a name and is used as a prefix

Element	oxygen	sulfur	selenium	nitrogen	phosph orous	silicon	boron
Valence	II	II II		III	Ш	IV	III
Prefix	Oxa	Thia	Selena	Aza	Phospha	Sila	Bora

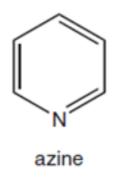
- If there are more than one heteroatoms in a ring, the **priority order** is from left to right in the table
- The "a" ending is dropped if the next syllable starts with a vowel example: "aza-irine" is properly written "azirine"

Rule #2: Ring size is designated by stems that follow the prefix:

Ring Size	3	4	5	6	7	8	9
Stems	-ir-	-et-	-ol-	-in-	-ep-	-oc-	-on-

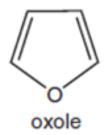
Rule #3 If the ring is fully unsaturated, the name is concluded with a suffix for ring size:

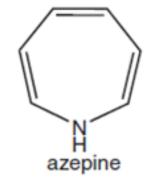
- 3 atoms, -ene (except -ine for N)
- 4, 5, and 6 atoms, -e
- 7, 8, and 9 atoms, -ine







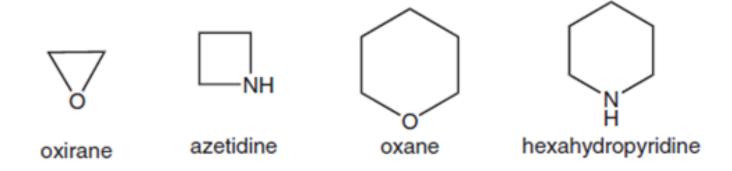




Rule #4:

If **fully saturated**, the name is concluded with a **suffix** for ring size:

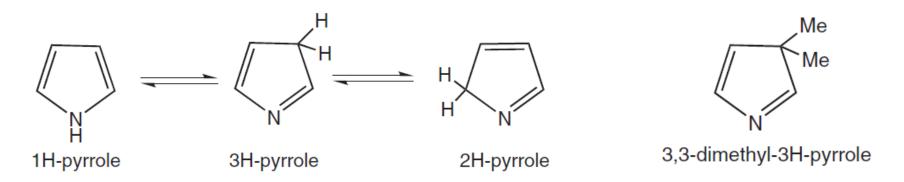
- the suffix is -ane for all ring sizes,
- except for N which uses -idine for rings of 3, 4, or 5 atoms, and for 6 atoms, a prefix of hexahydro- is used.
- the name oxane, not oxinane, is used for the 6-membered ring with O present



Exceptions: there are several exceptions and modifications to accommodate conflicts with prior usage!

The Extra Hydrogen

- There is a special problem resulting from isomerism in certain heterocyclic systems that requires clarification in the name
- simply adding a prefix that indicates the number of the ring atom that possesses the hydrogen, thus, 1H, 2H, and 3H
- Consider the case of pyrrole: There are actually two additional isomeric forms that result from apparent 1,3-shifts of hydrogen



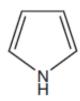
 The extra-hydrogen problem can occur in any ring system of nitrogen containing an odd number of ring atoms or oxygen containing even number of ring atoms

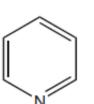
Examples

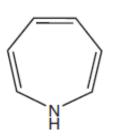


















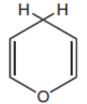


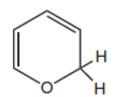
B. Oxygen Heterocyclic Parents

















C. Sulfur Heterocyclic Parents







