2302687 – Heterocyclic Compounds – Part I

Lecture 1-1



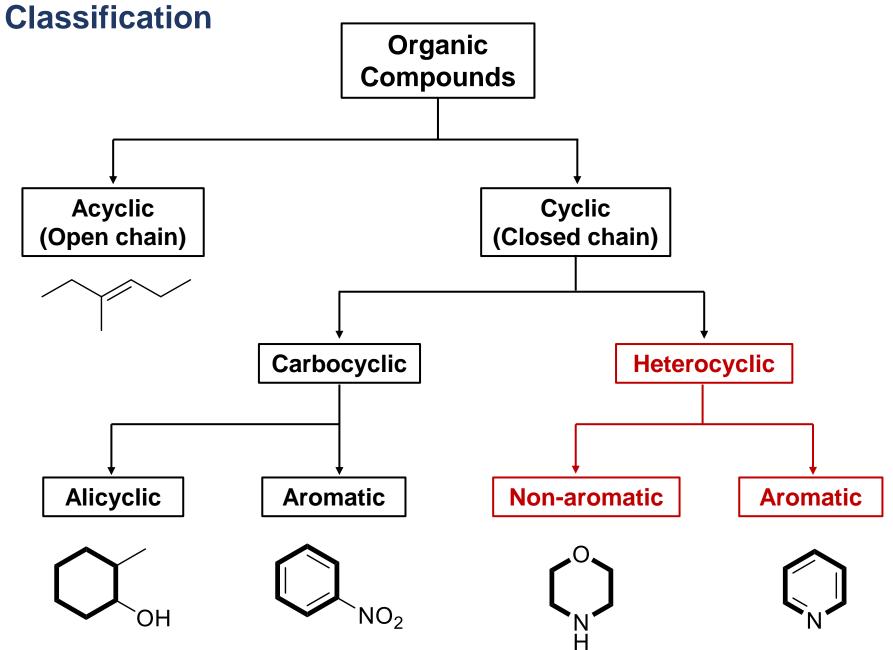
Introduction to Heterocyclic Compounds



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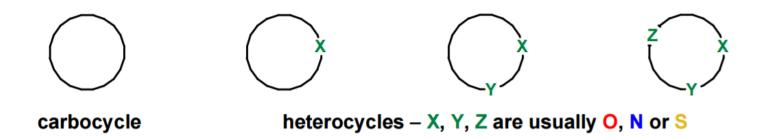
Recommended Textbook:

Heterocyclic Chemistry, 5th Edition, J. A. Joule, K. Mills, **2010**, Wiley



Heterocyclic Compounds

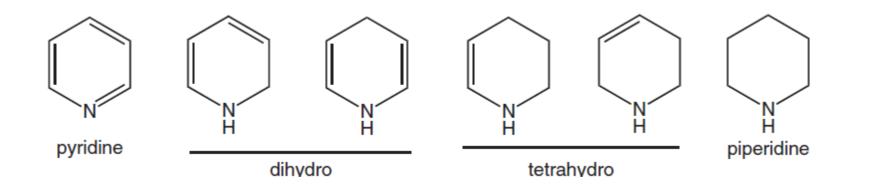
 Heterosubstituted rings are those in which one or more carbon atoms in a purely carbon-containing ring (known as a carbocyclic ring) is replaced by some other atom (referred to as a heteroatom).



- In practice, the most commonly found heteroatom is nitrogen, followed by oxygen and sulphur
- However, many other atoms can form the stable covalent bonds; of note are phosphorus, arsenic, antimony, silicon, selenium, tellurium, boron, and germanium

Heterocyclic Compounds

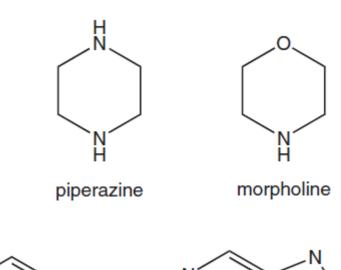
 Pyridine is an excellent example of a simple heterocycle. Here, one carbon of benzene is replaced by nitrogen, without interrupting the classic unsaturation and aromaticity



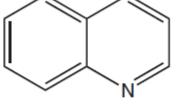
- Similarly, replacement of a carbon in cyclohexane by nitrogen produces the saturated heterocycle piperidine
- Between these extremes of saturation come several structures with one or two double bonds

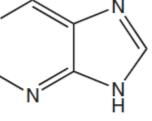
Heterocyclic Compounds

 Rings may have more than one heteroatom, which may be the same or different, as in the examples that follow



 To broaden the field, other rings may be **fused** onto a parent heterocycle





quinoline

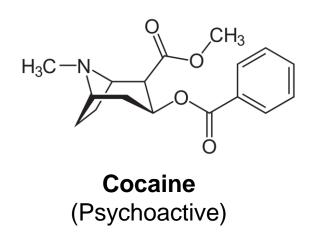
purine

- 133,326 different heterocyclic ring systems had been reported by 1984, and many more have been reported since then
- Moreover, hydrogens on these rings can be replaced by a multitude of substituents. As a result, millions of heterocyclic compounds are known! (24,282,284 compounds in 2007 !)

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Heterocyclic Compounds in Nature

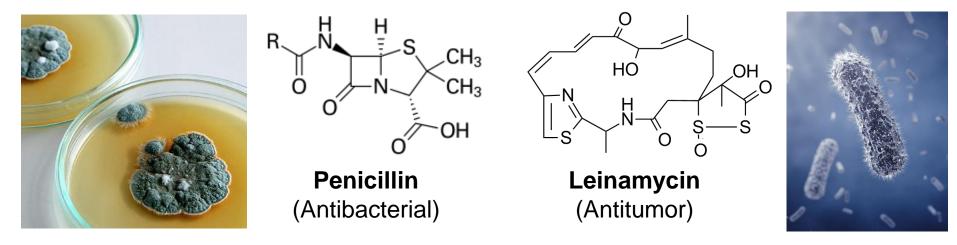
- Nature abounds in heterocyclic compounds, many of profound importance in biological processes
- We find heterocyclic rings in **vitamins**, **coenzymes**, **porphyrins** (like hemoglobin), **DNA**, **RNA**, and so on
- The **plant** kingdom contains thousands of nitrogen heterocyclic compounds, most of which are weakly basic and called **alkaloids**



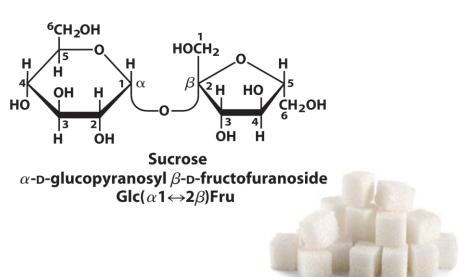


Heterocyclic Compounds in Nature

 Complex heterocyclic compounds are elaborated by microorganisms and are useful as antibiotics in medicine



 The huge field of carbohydrate chemistry depends on heterocyclic frameworks; all disaccharides and polysaccharides have rings usually of five (called furanose) or six (called pyranose) members that contain an oxygen atom



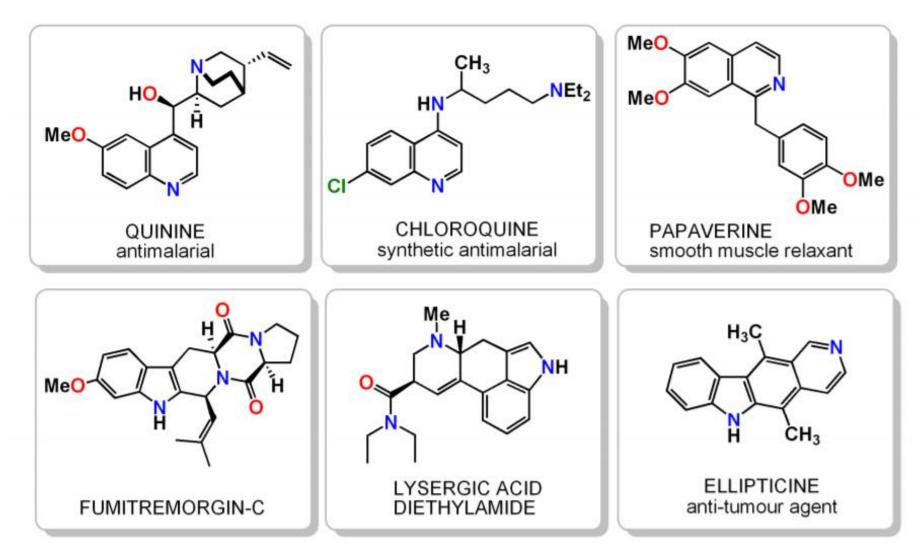
Applications of Heterocyclic Compounds

- Heterocyclic compounds can be synthesized in many ways
- Many synthetic (as well as natural) heterocyclic compounds are of extreme value as medicinals, agrochemicals, plastics precursors, dyes, photographic chemicals, and so on, and new structures are constantly being sought in research in these areas



Applications of Heterocyclic Compounds

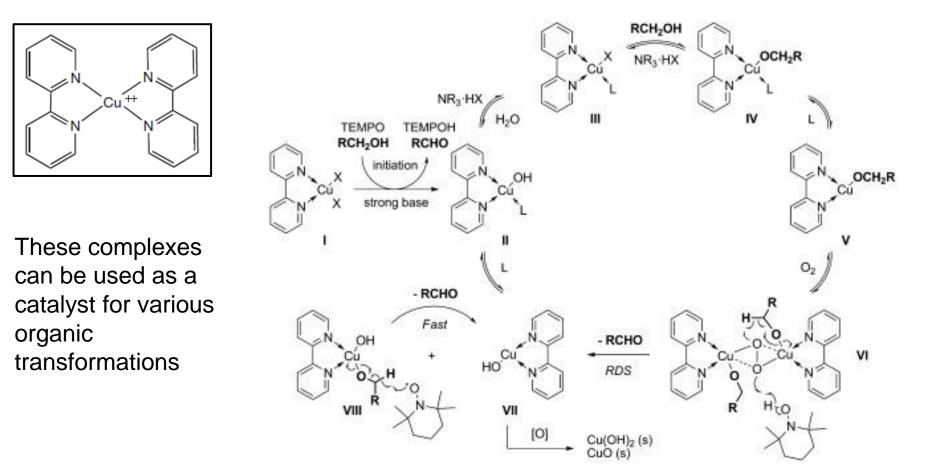
Medicinal chemistry especially is associated intimately with heterocyclic compounds; most of all chemicals used in medicine are based on heterocyclic frameworks



Applications of Heterocyclic Compounds

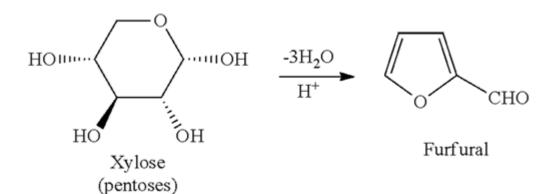
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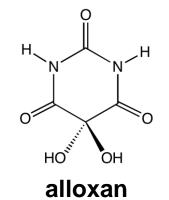
- Heterocyclic compounds can be useful **ligands** (through their lone pair electrons or pi-systems) in the construction of coordination complexes
- An example of a heterocycle frequently used for this purpose is 2,2-bipyridyl, which is shown here as complexed to cupric ion



History of Heterocyclic Compounds

• 1818: Brugnatelli isolates alloxan from uric acid

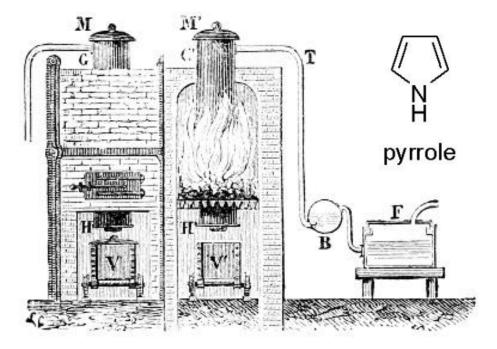




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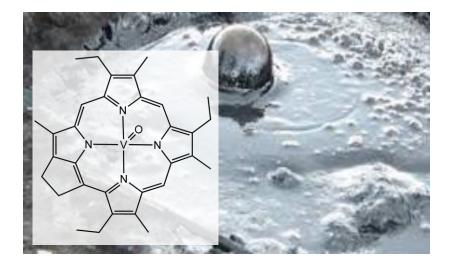
 1832: Dobereiner produces furfural by treating starch with sulfuric acid

 1834: Runge obtains pyrrole ("fiery oil") by dry distillation of bones

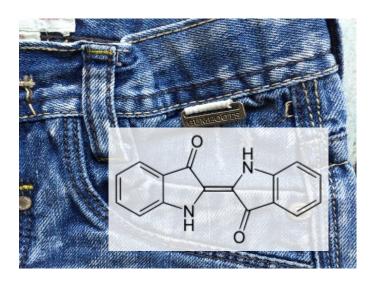


History of Heterocyclic Compounds

• **1906**: Friedlander synthesizes **indigo dye**, allowing synthetic chemistry to displace a large agricultural industry



 1951: Chargaff's rules are described, highlighting the role of heterocyclic compounds in the genetic code (purines and pyrimidines)



 1936: Treibs isolates chlorophyl derivatives from crude oil, explaining the biological origin of petroleum

