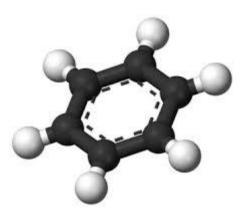


Aromatic – Coupling Reactions



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Recommended Textbook:

Chapter 17 in *Organic Chemistry*, 8th Edition, L. G. Wade, Jr., **2010**, Prentice Hall (Pearson Education)

Coupling Reactions

 Organic chemists have developed new coupling reactions (reactions that form C-C bonds) using a wide variety of methods that tolerate many other functional groups.

- Some of the most successful coupling reactions use **transition metals** that change valences easily, adding and eliminating substituents as they pass from one oxidation state to another.
 - Copper (Organocuprate Reagents)
 - Palladium
 - Heck Reaction
 - Suzuki Reaction

Couplings Using Organocuprate Reagents

 Lithium dialkylcuprate reagents (Gilman reagents) are formed by the reaction of two equivalents of an organolithium reagent with cuprous iodide.

 $\begin{array}{l} R - X \\ (R = alkyl, vinyl, aryl) \end{array}$

• The organocuprate can be coupled with an alkyl, vinyl, or aryl halide

R₂CuLi

Example



iodobenzene lithium divinylcuprate

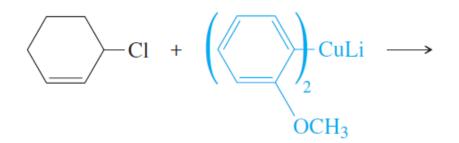
Couplings Using Organocuprate Reagents

More Examples

 A vinyl halide with an aryl cuprate, preserving the stereochemistry of the vinyl halide



• An alkyl halide with an aryl cuprate.



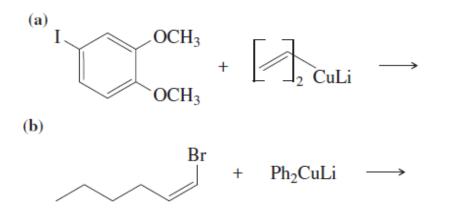
• An acyl halide with an organocuprate, giving a ketone

$$\bigcirc O \\ Cl + (\bigcirc 2) CuLi \longrightarrow$$

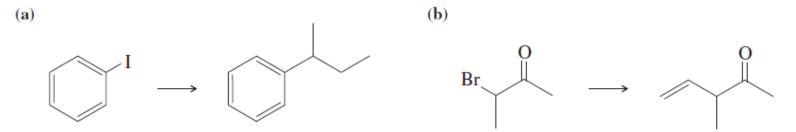
Couplings Using Organocuprate Reagents

Problems

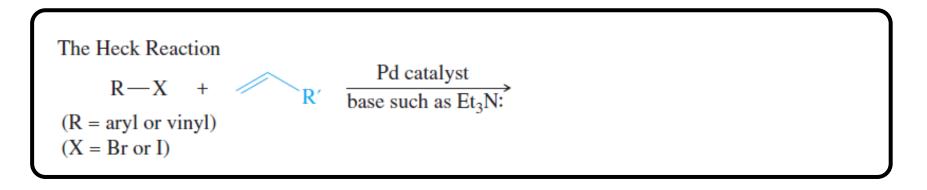
What products would you expect from the following reactions?



What organocuprate reagent would you use for the following substitutions?



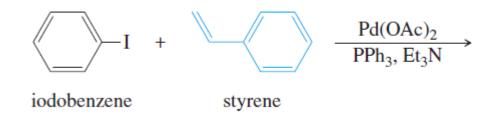
 The Heck reaction is the palladium-catalyzed coupling of an aryl or vinyl halide with an alkene to give a new C—C bond at the less substituted end of the alkene, usually with trans stereochemistry



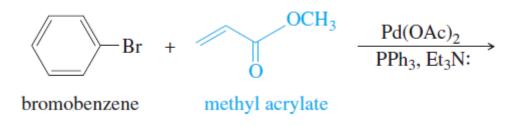
- In most cases, the **halide** is a bromide or an iodide
- the **alkene** is typically monosubstituted.
- The **palladium** catalyst might be Pd(OAc)₂ or PdCl₂ or a variety of other palladium compounds.
- A **base** such as triethylamine or sodium acetate is added to neutralize the HX released in the reaction
- Many reactions use **triphenylphosphine (PPh₃)** to complex with the palladium, which helps stabilize it and enhances its reactivity

Examples

• An aryl halide with an aryl olefin.



• An aryl halide with a conjugated acid or ester.



• A vinyl halide with a conjugated nitrile.

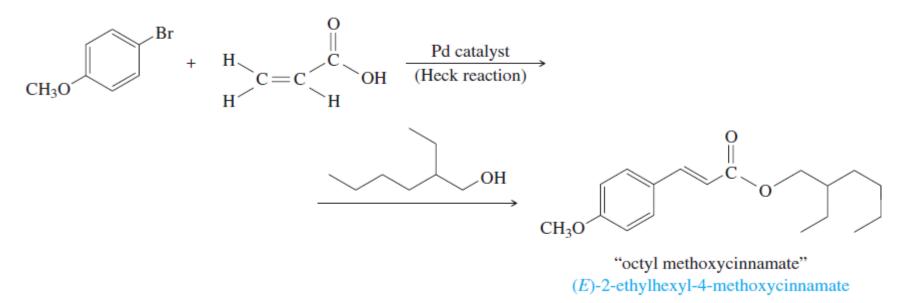
$$(E)-1-iodobut-I-ene acrylonitrile \xrightarrow{Pd(OAc)_2}{PPh_3, Et_3N:}$$

More examples

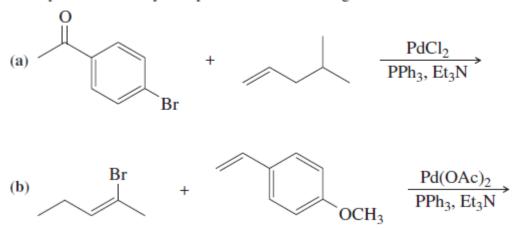
 The industrial synthesis of naproxen, an over-the-counter analgesic and antiinflammatory drug.



• The industrial synthesis of octyl methoxycinnamate, a common ingredient in sunscreens.

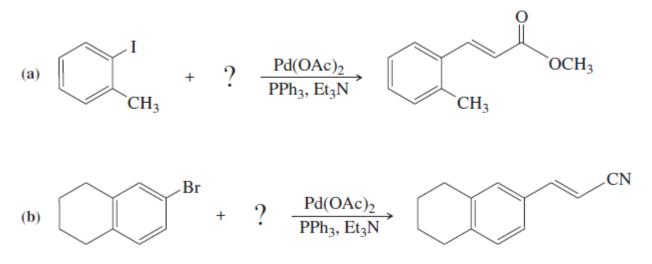


Problems



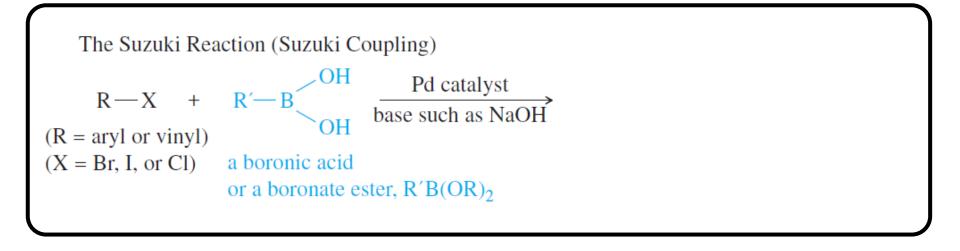
What products would you expect from the following reactions?

What substituted alkene would you use in the Heck reaction to make the following products?



The Suzuki Reaction

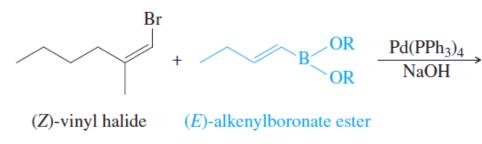
 The Suzuki reaction is a palladium-catalyzed substitution that couples an aryl or vinyl halide with an alkyl, alkenyl, or aryl boronic acid or boronate ester.



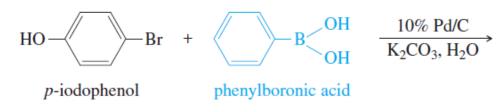
 Like the Heck reaction, the Suzuki coupling can use water as the solvent.
Waterbased Suzuki reactions are attractive both for industrial processes and for labs that want to minimize the purchase and disposal of toxic solvents.

The Suzuki Reaction Examples

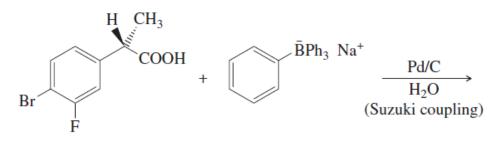
• A vinyl halide with an alkenylboronate ester, preserving the stereochemistry of the reagents.



• An aryl halide with an arylboronic acid, using palladium on carbon and water as the solvent.



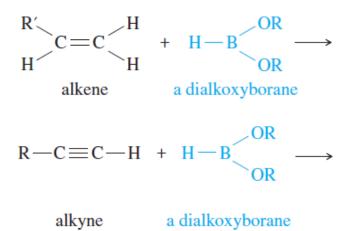
• Synthesis of the anti-inflammatory drug flurbiprofen by an environmentally efficient Suzuki coupling that uses water as the solvent and palladium on carbon as a reusable catalyst.



sodium tetraphenylborate

The Suzuki Reaction

 Alkyl and vinyl boronate esters are synthesized by the hydroboration of double and triple bonds, similar to the hydroboration of alkenes and alkynes



 Arylboronate esters are made by a two-step process: First, we convert the aryl halide to the aryllithium compound. Addition of a trialkyl borate (often trimethyl borate) allows the organolithium compound to form a carbon-boron bond and expel an alkoxide group.

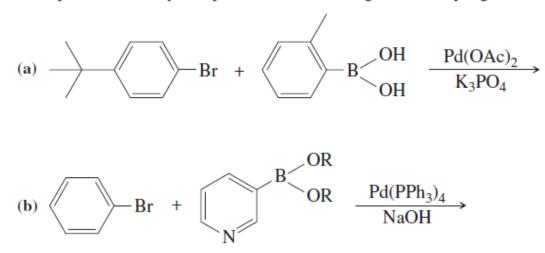
$$Br \xrightarrow{2 \text{ Li}}$$

B(OMe)₃ (trimethyl borate)

bromobenzene

The Suzuki Reaction Problems

What products would you expect from the following Suzuki coupling reactions?



Show how you would use Suzuki reactions to synthesize these products from the indicated starting materials. You may use any additional reagents you need.

