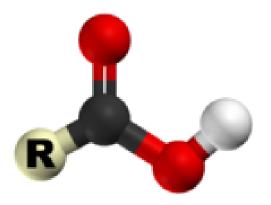
# 2302106 – Basic Organic Chemistry for ISE – Part II Lecture 5-4

#### **Carboxylic and Derivatives – Nucleophilic Substitution-2**



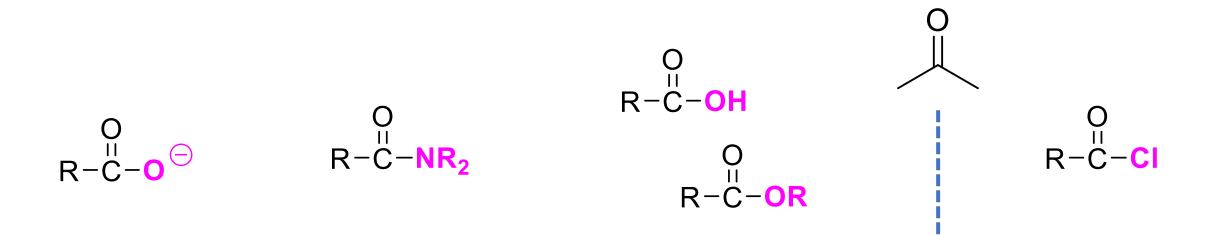
Instructor: Asst. Prof. Dr. Tanatorn Khotavivattana E-mail: tanatorn.k@chula.ac.th

**Recommended Textbook:** 

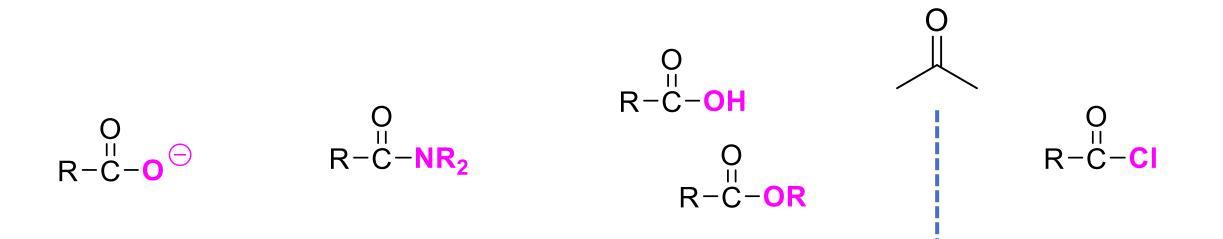
Chapter 20 in Organic Chemistry, 8th Edition, L. G. Wade, Jr., 2010, Prentice Hall (Pearson Education)

1

1) Reduction with Hydride sources: Sodium borohydride (NaBH<sub>4</sub>)



1) Reduction with Hydride sources: Lithium aluminum hydride (LiAlH<sub>4</sub>)



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**Esters to alcohols** 

O II R-C-OR

1) Reduction with Hydride sources: Lithium aluminum hydride (LiAlH<sub>4</sub>)

**Carboxylic acids to alcohols** 

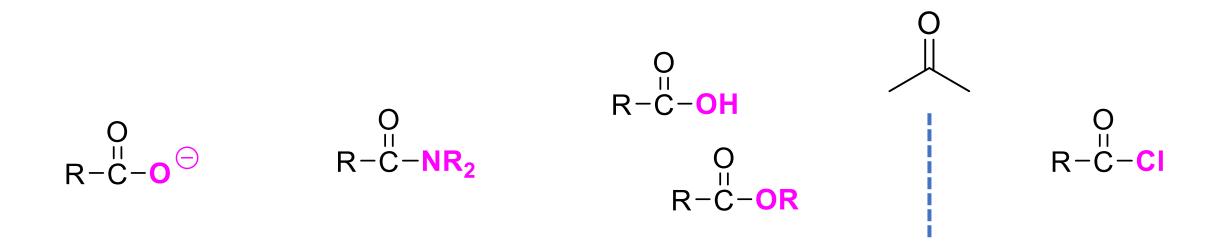
O R-C-OH

1) Reduction with Hydride sources: Lithium aluminum hydride (LiAlH<sub>4</sub>)

Amides to amines

O II R-C-NR<sub>2</sub>

#### 2) Alkylation with alkyl lithium



2) Alkylation with alkyl lithium

Acid chlorides / Esters to alcohols

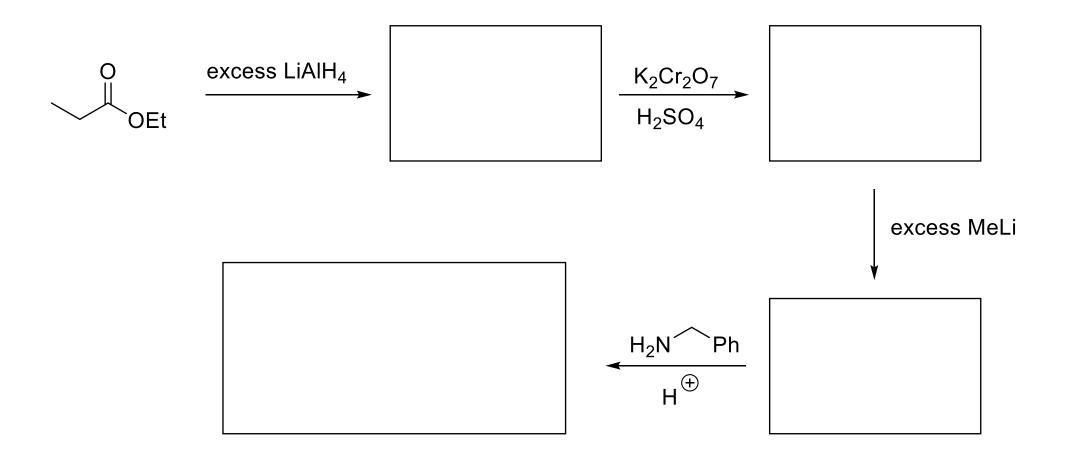
O II R-C-OR

2) Alkylation with alkyl lithium

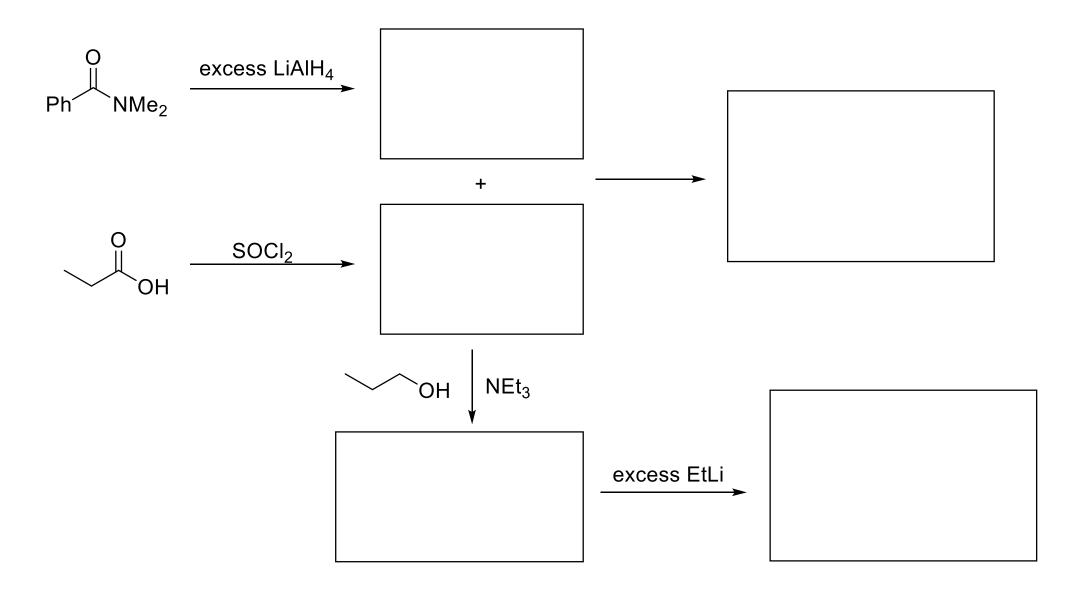
**Carboxylic acids to ketones** 

O R-C-OH

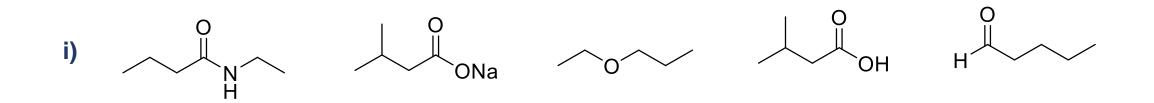
**Example: Fill the gap in the following scheme** 



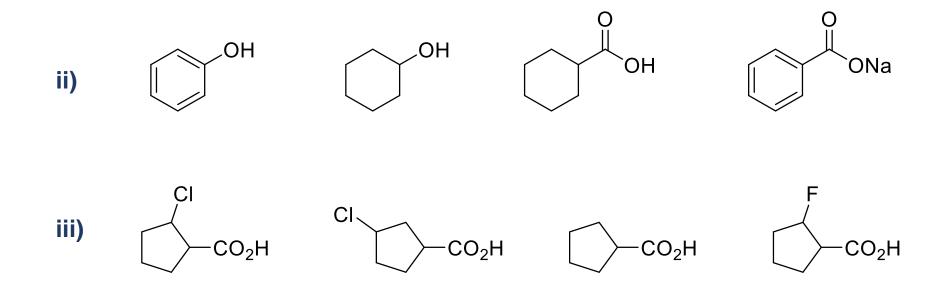
#### **Example: Fill the gap in the following scheme**



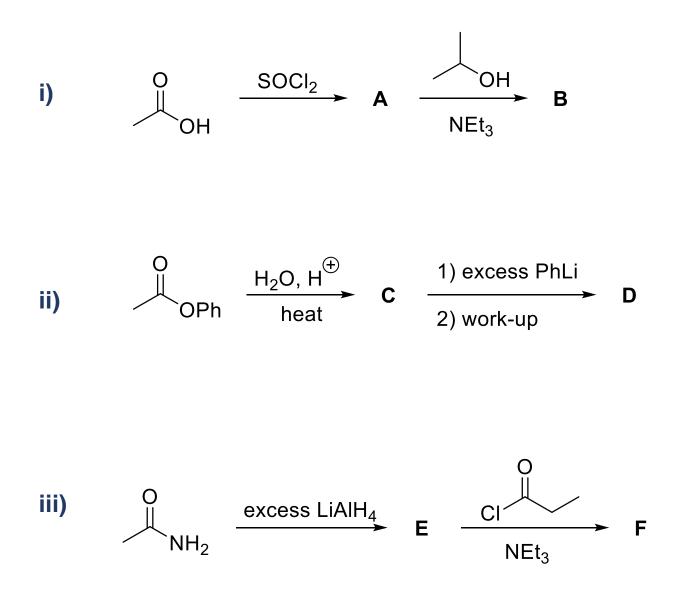
Arrange each group of compounds in order of increasing boiling point



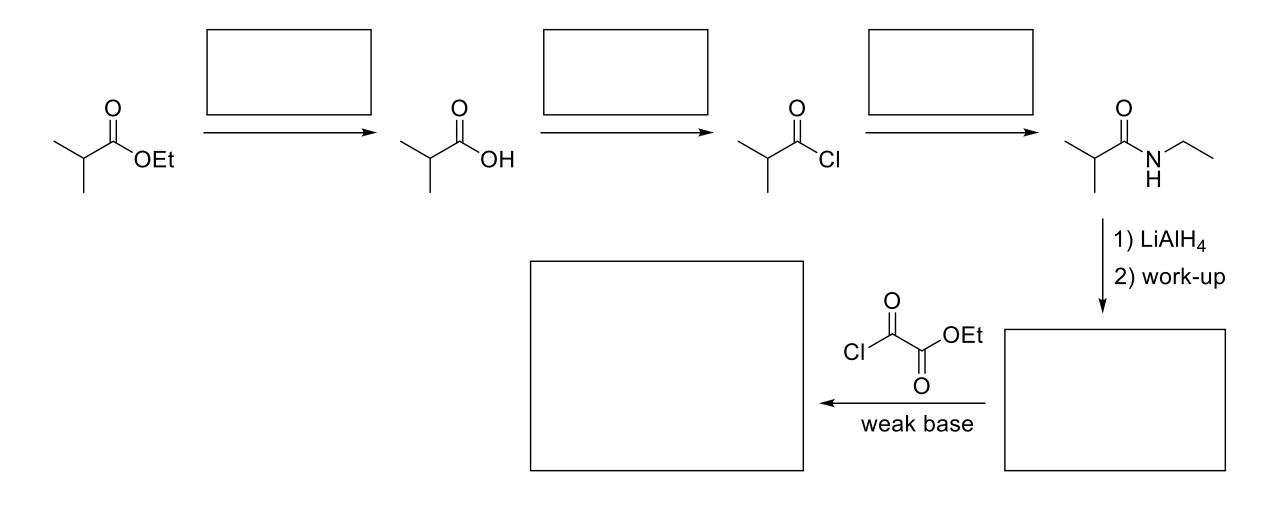
Arrange each group of compounds in order of increasing acidity



Predict the product and draw the mechanism of the following reactions



#### Fill the gap in the following scheme



#### Fill the gap in the following scheme

