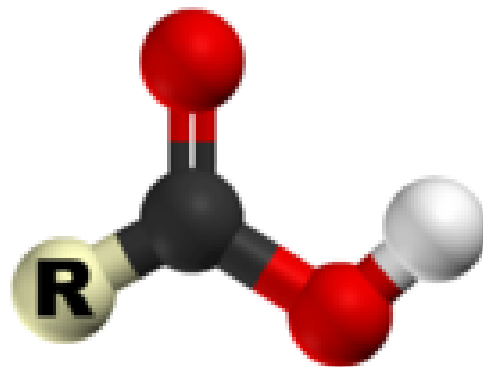


Carboxylic and Derivatives – Nucleophilic Substitution-1

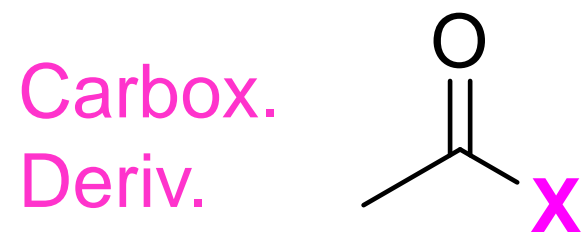
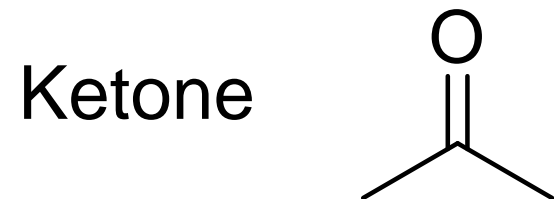


Instructor: Asst. Prof. Dr. Tanatorn Khotavivattana
E-mail: tanatorn.k@chula.ac.th



Recommended Textbook:

Chapter 20 in *Organic Chemistry*, 8th Edition, L. G. Wade, Jr., **2010**, Prentice Hall (Pearson Education)

Nucleophilic Addition vs. Substitution



Nucleophilic Substitution – Reactivity towards Nucleophile

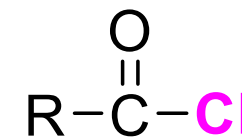
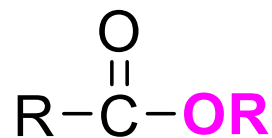
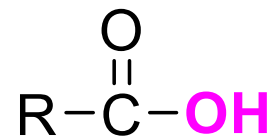
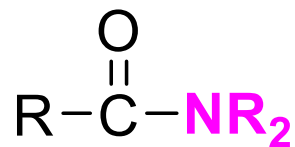
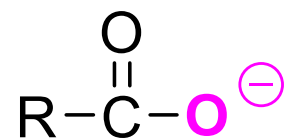
<i>Reactivity</i>	<i>Derivative</i>	<i>Leaving group</i>	<i>Basicity</i>
more reactive 	acid chloride $\text{R}-\overset{\text{O}}{\parallel}{\text{C}}-\text{Cl}$	Cl^-	less basic 
	ester $\text{R}-\overset{\text{O}}{\parallel}{\text{C}}-\text{O}-\text{R}'$	$-\text{O}-\text{R}'$	
	amide $\text{R}-\overset{\text{O}}{\parallel}{\text{C}}-\text{NH}_2$	$-\text{NH}_2$	
less reactive	carboxylate $\text{R}-\overset{\text{O}}{\parallel}{\text{C}}-\text{O}^-$	$-$	more basic

Factor #1 – Leaving group ability

Factor #2 – Resonance stabilization

Nucleophilic Substitution – Reaction with Weak Nucleophiles

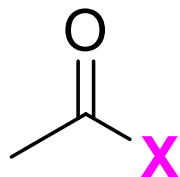
1 – Reaction with **water** (no catalyst)



Nucleophilic Substitution – no catalyst vs. base/acid catalyzed

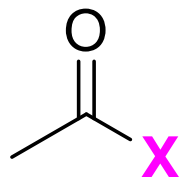
#1 No catalyst

Nu-H



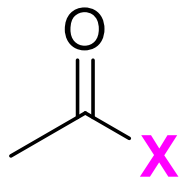
#2 Base catalyzed

Nu-H



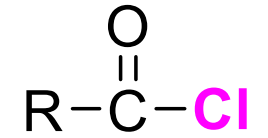
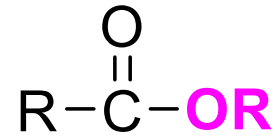
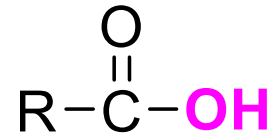
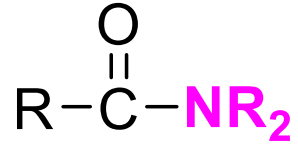
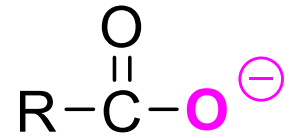
#3 Acid catalyzed

Nu-H



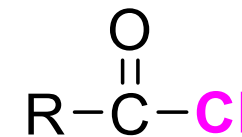
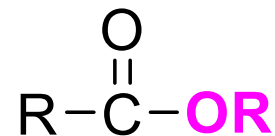
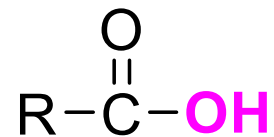
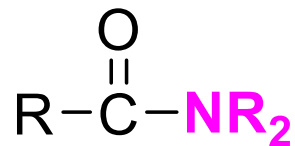
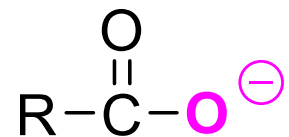
Nucleophilic Substitution

1 – Reaction with **water** (base catalyzed)



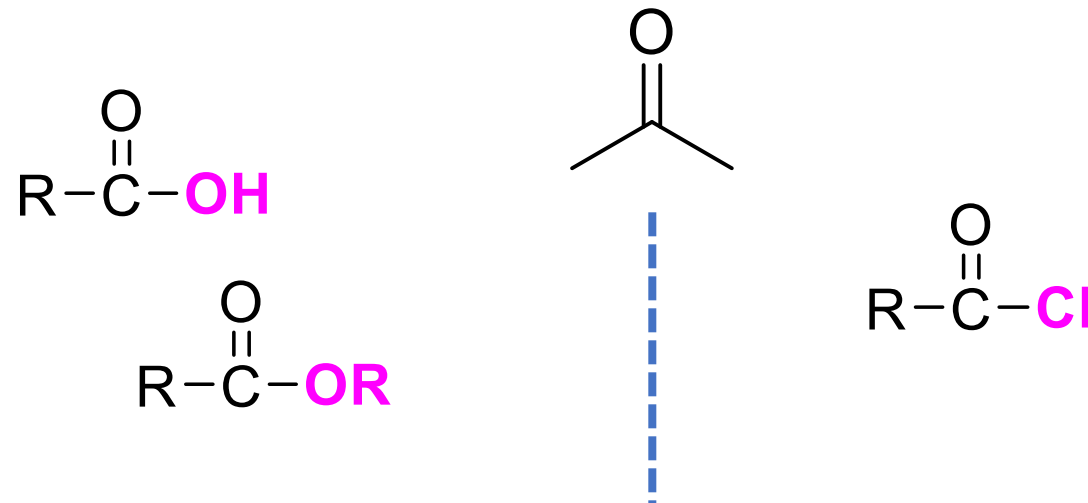
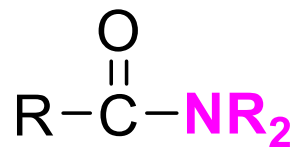
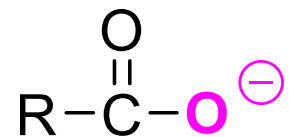
Nucleophilic Substitution

1 – Reaction with **water** (acid catalyzed)



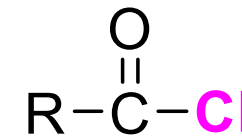
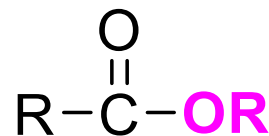
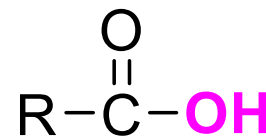
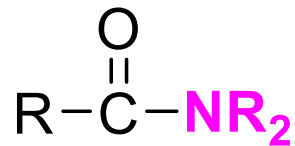
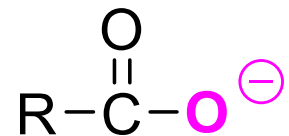
Nucleophilic Substitution

2 – Reaction with alcohol (no catalyst)



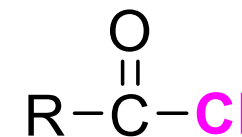
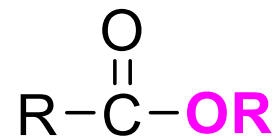
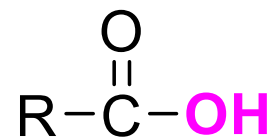
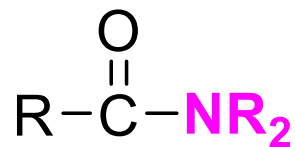
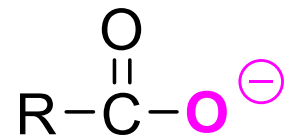
Nucleophilic Substitution

2 – Reaction with alcohol (base catalyzed)



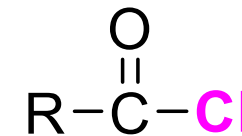
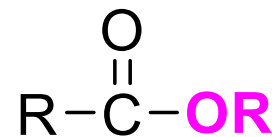
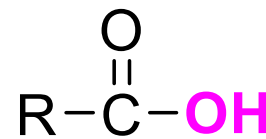
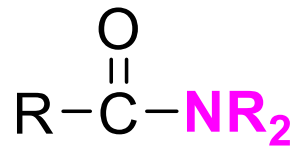
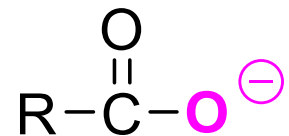
Nucleophilic Substitution

2 – Reaction with alcohol (acid catalyzed)



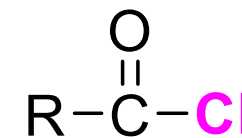
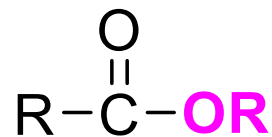
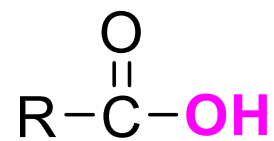
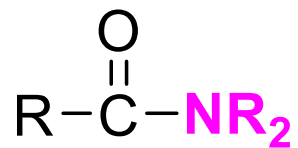
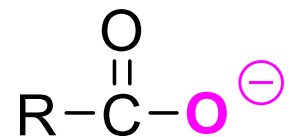
Nucleophilic Substitution

3 – Reaction with **amine** (no catalyst)



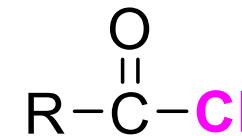
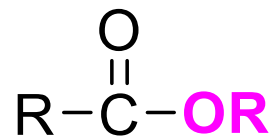
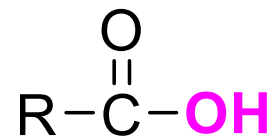
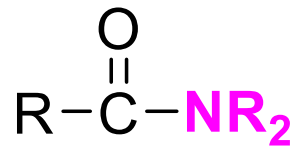
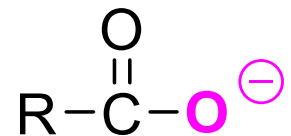
Nucleophilic Substitution

3 – Reaction with **amine** (base catalyst)



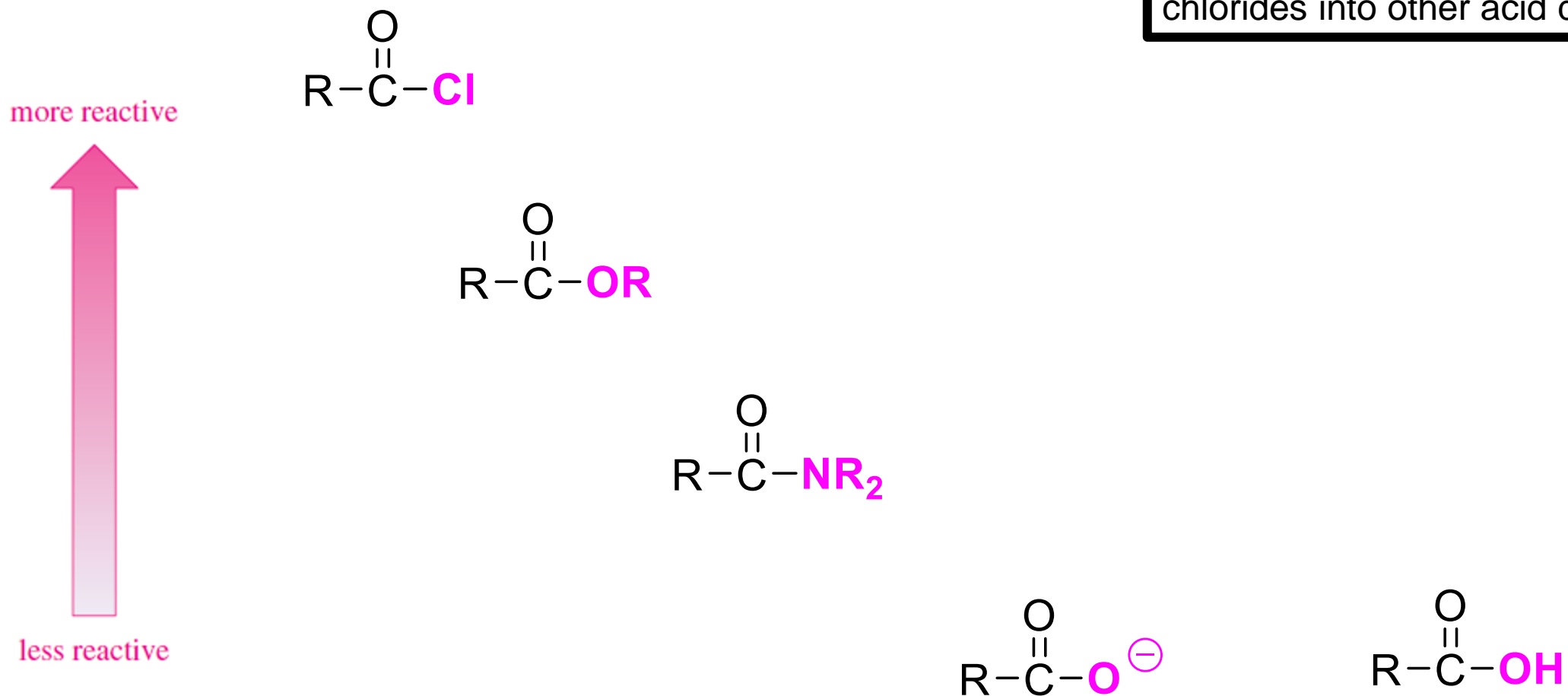
Nucleophilic Substitution

3 – Reaction with **amine** (acid catalyst)



Favourable Interconversion of Acid Derivatives

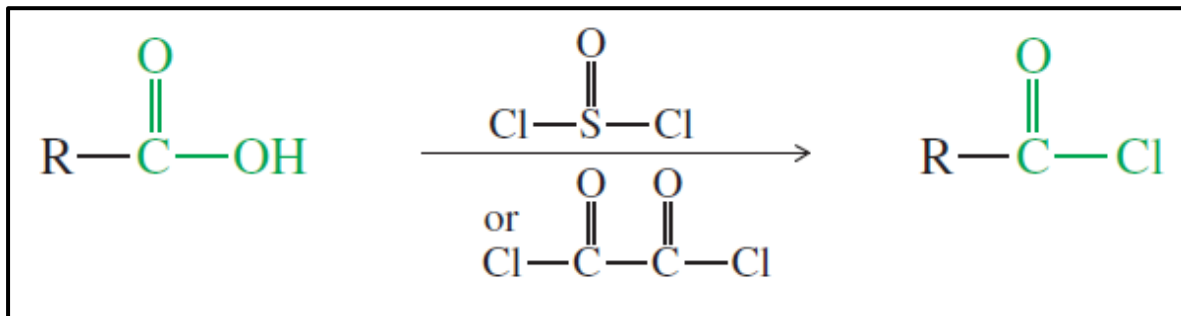
- Acid chlorides are the **most reactive** acid derivatives



The most frequent strategy is to convert carboxylic acids into acid chlorides using **thionyl chloride (SOCl₂)**; then convert the acid chlorides into other acid derivatives

Synthesis of Acid Chlorides

The best reagents for converting carboxylic acids to acid chlorides are **thionyl chloride (SOCl₂)** and **oxalyl chloride [(COCl)₂]** because they form **gaseous by-products** that do not contaminate the product.



Mechanism:

Nucleophilic Substitution - Example

Fill the gap in the following scheme

